

# Motivations for Quantum Mechanics

- Atomic spectra (emission spectra, absorption spectra, Balmer series, Rydberg relationship, Bohr's semi-classical theory, Bohr-Sommerfeld theory)
- Photoelectric effect (emission from metals, work function, dependence on wavelength, dependence on intensity)

# Motivations for Quantum Mechanics

- Rayleigh-Jeans radiation theory
- Calculation using classical oscillators in equilibrium with electromagnetic radiation
- Jeans corrected Rayleigh's theory with an additional factor of 2 for polarization
- Theory predicted infinite intensity at infinite frequency—“ultraviolet catastrophe”

# Major Steps in Formulating Quantum Mechanics

- Einstein proposed a theory of the photoelectric effect in which light energy was quantized as “photons” with energy  $h\nu$ , where  $h$  is a constant ( $6.6 \times 10^{-34}$  J-s) and  $\nu$  is the frequency of the radiation
- Planck proposed modification of Rayleigh calculation with oscillators with quantized in steps of  $h\nu$ .  $h$  is now called Planck’s constant.

# Major Steps in Formulating Quantum Mechanics

- deBroglie proposes that particles are characterized by a wavelength  $\lambda = h/p$ . Bohr quantization then results from a circular boundary condition
- Schroedinger proposes that all mechanics can be described in terms of a wave equation—a partial differential equation in space and time

# Major Steps in Formulating Quantum Mechanics

- Simultaneously, Heisenberg formulates an equivalent theory of quantum mechanics based on operator-algebra and vector spaces

# Schroedinger Equation

$$-\frac{\hbar^2}{2m} \frac{\partial^2 \Psi(x,t)}{\partial x^2} + U(x)\Psi(x,t) = i\hbar \frac{\partial \Psi(x,t)}{\partial t}$$

$\hbar = \frac{h}{2\pi}$ ,  $\Psi(x,t)$  is called the wave function

$$P(x,t) = |\Psi(x,t)|^2 = \Psi^* \Psi$$

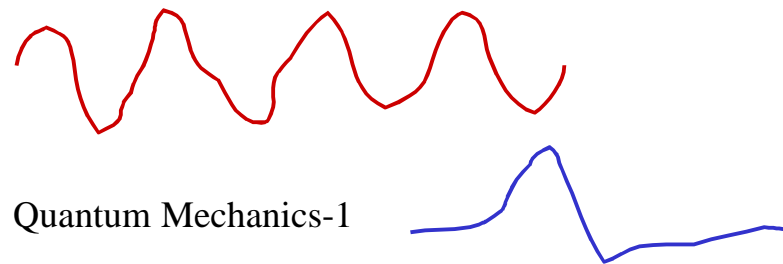
$$\int_{\text{all space}} \Psi^* \Psi = 1$$

# Schroedinger Equation

- Premise of quantum mechanics is that all knowable information about a system is contained in the wave function
- Note that a particle wave function does not say where a particle is. It only gives the probability that a particle is at a specific location

# Heisenberg Uncertainty Principle

- Schroedinger formulation of quantum mechanics implies the Heisenberg uncertainty principle
- Uncertainty principle exists for all waves
- A monochromatic wave is spatially diffuse
- A compact wave packet consists of many frequencies



# Wave Function

- Since all knowable information is contained in the wave function, everything knowable can be determined by solving the Schroedinger equation for the wave function
- A general approach to solving PDE's if U depends on x alone is a separation of variables

$$\Psi(x, t) = \varphi(x)e^{-i\omega t}$$

# Time-Independent SE

$$-\frac{\hbar^2}{2m} \frac{\partial^2 \Psi(x,t)}{\partial x^2} + U(x)\Psi(x,t) = i\hbar \frac{\partial \Psi(x,t)}{\partial t}$$

$$\Psi(x,t) = \varphi(x)e^{-i\omega t}$$

$$-\frac{\hbar^2}{2m} \frac{\partial^2 \varphi(x)e^{-i\omega t}}{\partial x^2} + U(x)\varphi(x)e^{-i\omega t} = (i\hbar)(-i\omega)\varphi(x)e^{-i\omega t}$$

$$-\frac{\hbar^2}{2m} \frac{\partial^2 \varphi(x)}{\partial x^2} + U(x)\varphi(x) = \hbar\omega\varphi(x) = E\varphi(x)$$

$$H = -\frac{\hbar^2}{2m} \frac{\partial^2}{\partial x^2} + U(x), \quad H\varphi(x) = E\varphi(x)$$

# Time-Independent SE

- If the potential  $U(x,t)$  is a function of  $x$  alone (*i.e.*, time independent), then the solution of the SE (*i.e.*, the wave function) can be written as the product of

$$\varphi(x) \text{ and } e^{-i\omega t}$$

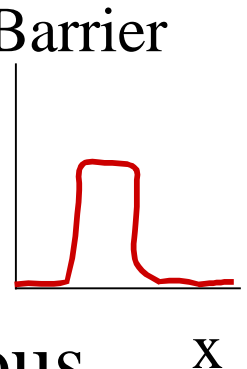
# Time-Independent SE

- The spatial term  $\psi(x)$  is a solution of a total differential equation known as the time-independent SE
- The time-independent SE can be written as shown on the next slide where  $H$  is an operator called the Hamiltonian
- The energies  $E$  are the eigenvalues of  $H$

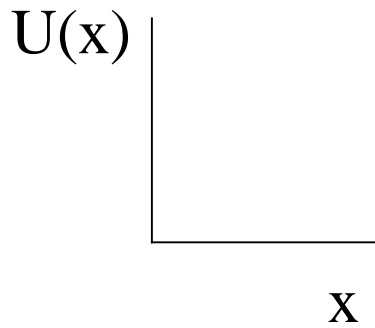
# Time-Independent SE

$$H = -\frac{\hbar^2}{2m} \frac{\partial^2}{\partial x^2} + U(x), \quad H\varphi(x) = E\varphi(x)$$

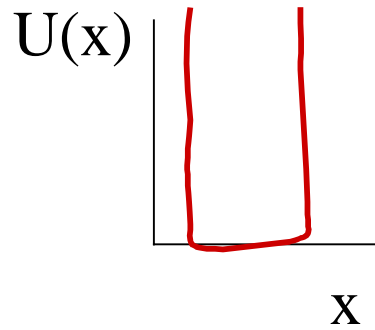
# Time-Independent SE $U(x)$ Barrier



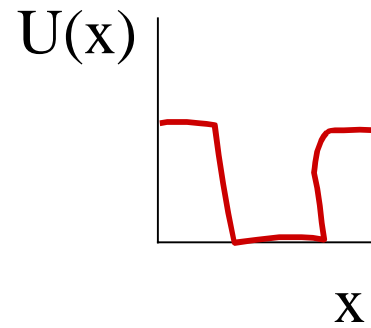
- Solve the time-independent SE for various potential energy functions



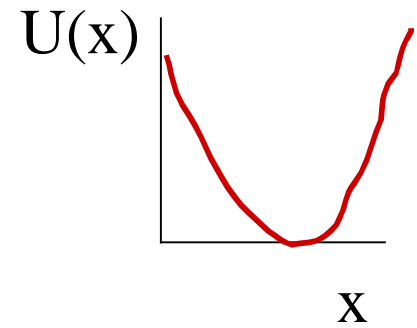
Free particle



Infinite Square Well



Finite Square Well



Harmonic Oscillator

# Time-Independent SE

- Solution to time-independent SE in 3-dimensions uses Spherical Harmonics (generalization of Legendre Polynomials)
- Identical particles requires symmetric (anti-symmetric) wave function depending on whether particles are bosons (fermions)